

Discussion Class 7

19 October 2007

1. The *induced dipole moment* of a neutral atom is approximately proportional to the external electric field:

$$\mathbf{p} = \alpha \mathbf{E}$$

The constant of proportionality α is called **atomic polarizability**.

Now consider a hydrogen atom in its ground state. The electron cloud has a charge density

$$\rho(r) = \frac{q}{\pi a^3} e^{-2r/a},$$

where q is the charge of the electron and a is the Bohr radius ($\equiv 0.5 \times 10^{-10}$ m). Find the atomic polarizability of such an atom. (For comparison, the experimental value is $0.66 \times 10^{-30} \text{m}^3$.)

[Hint: First calculate the electric field of the electron cloud, $E_e(r)$. The nucleus will be shifted from $r = 0$ to its equilibrium position d where $E_e = E_{\text{external}}$; expand the exponential in E_{external} , assuming $d \ll a$ and neglecting the higher order terms.]

2. The polarization in a linear dielectric is proportional to the field:

$$\mathbf{P} = \epsilon_0 \chi_e \mathbf{E} \tag{1}$$

If the material consists of atoms (or nonpolar molecules), the induced dipole moment of each atom is likewise proportional to the field:

$$\mathbf{p} = \alpha \mathbf{E} \tag{2}$$

What is the relation between the atomic polarizability α and the susceptibility χ_e ?

[Hint: Since \mathbf{P} (dipole moment per unit volume) is \mathbf{p} (dipole moment per atom) times N (number of atoms per unit volume), the obvious guess will be

$$\epsilon_0 \chi_e = N \alpha \tag{3}$$

However, there is a subtle problem, for the field \mathbf{E} in Eq. 1 is the *total macroscopic field* in the medium, whereas the field in Eq. 2 is due to everything *except* the particular atom under consideration (by definition of atomic polarizability; see Problem 1); denote this field as \mathbf{E}_{else} . Assuming each atom to be a sphere of radius R , show that

$$\mathbf{E} = \left(1 - \frac{N\alpha}{3\epsilon_0}\right) \mathbf{E}_{\text{else}} \tag{4}$$

Use this to obtain

$$\begin{aligned}\chi_e &= \frac{N\alpha/\epsilon_0}{1 - N\alpha/3\epsilon_0}, \\ \text{or } \alpha &= \frac{3\epsilon_0}{N} \left(\frac{\epsilon_r - 1}{\epsilon_r + 2} \right)\end{aligned}\tag{5}$$

Solution

1. We calculate the field at radius r using Gauss's law:

$$\oint \mathbf{E} \cdot d\mathbf{a} = \frac{Q_{\text{enc}}}{\epsilon_0}, \text{ or } E = \frac{Q_{\text{enc}}}{4\pi\epsilon_0 r^2}$$

The total charge enclosed within a radius r is given by

$$\begin{aligned}Q_{\text{enc}} &= \int_0^r \rho d\tau = \frac{4\pi q}{\pi a^3} \int_0^r e^{-2\tilde{r}/a} \tilde{r}^2 d\tilde{r} \\ &= \frac{4q}{a^3} \left[-\frac{a}{2} e^{-2\tilde{r}/a} \left(\tilde{r}^2 + a\tilde{r} + \frac{a^2}{2} \right) \right]_0^r \\ &= q \left[1 - e^{-2r/a} \left(1 + \frac{2r}{a} + \frac{2r^2}{a^2} \right) \right].\end{aligned}$$

(Note that $Q_{\text{enc}}(r \rightarrow \infty) = q$.) So the field of the electron cloud is

$$E_e(r) = \frac{q}{4\pi\epsilon_0 r^2} \left[1 - e^{-2r/a} \left(1 + \frac{2r}{a} + \frac{2r^2}{a^2} \right) \right].$$

The proton in the nucleus will be shifted from $r = 0$ to the point d where $E_e = E$ (the external field):

$$E = \frac{q}{4\pi\epsilon_0 d^2} \left[1 - e^{-2d/a} \left(1 + \frac{2d}{a} + \frac{2d^2}{a^2} \right) \right].$$

Assuming $d \ll a$, we expand the exponential in powers of (d/a) :

$$\begin{aligned}e^{-2d/a} &= 1 - \left(\frac{2d}{a} \right) + \frac{1}{2} \left(\frac{2d}{a} \right)^2 - \frac{1}{3} \left(\frac{2d}{a} \right)^3 + \dots \\ &= 1 - \frac{2d}{a} + 2 \left(\frac{d}{a} \right)^2 - \frac{4}{3} \left(\frac{d}{a} \right)^3 + \dots \\ 1 - e^{-2d/a} \left(1 + \frac{2d}{a} + \frac{2d^2}{a^2} \right) &= 1 - \left(1 - \frac{2d}{a} + 2 \left(\frac{d}{a} \right)^2 - \frac{4}{3} \left(\frac{d}{a} \right)^3 + \dots \right) \left(1 + \frac{2d}{a} + \frac{2d^2}{a^2} \right) \\ &= \frac{4}{3} \left(\frac{d}{a} \right)^3 + \text{higher order terms}.\end{aligned}$$

So the external electric field is

$$E = \frac{q}{4\pi\epsilon_0 d^2} \left(\frac{4}{3} \frac{d^3}{a^3} \right) = \frac{p}{3\pi\epsilon_0 a^3}$$

with $p = qd$. So

$$\alpha = 3\pi\epsilon_0 a^3.$$

This result predicts $\frac{\alpha}{4\pi\epsilon_0} = \frac{3}{4}a^3 = \frac{3}{4}(0.5 \times 10^{-10})^3 = 0.09 \times 10^{-30}\text{m}^3$, compared to the experimental value $0.66 \times 10^{-30}\text{m}^3$. We can not expect a better result anyway in this *classical* picture. For a more sophisticated approach, see W.A. Bowers, *Am. J. Phys.* **54**, 347 (1986).

2. The density of atoms is $N = \frac{1}{(4/3)\pi R^3}$ assuming no voids in the material (idealized case). The macroscopic field \mathbf{E} is $\mathbf{E}_{\text{self}} + \mathbf{E}_{\text{else}}$, where \mathbf{E}_{self} is the average field over the sphere due to the atom itself. From Eq. 3.105,

$$\mathbf{E}_{\text{self}} = -\frac{1}{4\pi\epsilon_0} \frac{\mathbf{p}}{R^3}$$

Now $\mathbf{p} = \alpha \mathbf{E}_{\text{else}} \Rightarrow \mathbf{P} = N\alpha \mathbf{E}_{\text{else}}.$

So
$$\mathbf{E} = -\frac{1}{4\pi\epsilon_0} \frac{\alpha}{R^3} \mathbf{E}_{\text{else}} + \mathbf{E}_{\text{else}} = \left(1 - \frac{\alpha}{4\pi\epsilon_0 R^3} \right) \mathbf{E}_{\text{else}}$$

$$= \left(1 - \frac{N\alpha}{3\epsilon_0} \right) \mathbf{E}_{\text{else}}.$$

So
$$\mathbf{P} = \frac{N\alpha}{(1 - N\alpha/3\epsilon_0)} \mathbf{E} = \epsilon_0 \chi_e \mathbf{E},$$

and hence
$$\chi_e = \frac{N\alpha}{(1 - N\alpha/3\epsilon_0)}.$$

Solving for α :
$$\alpha = \frac{3\epsilon_0}{N} \frac{\chi_e}{3 + \chi_e}.$$

But $\chi_e = \epsilon_r - 1$, so

$$\alpha = \frac{3\epsilon_0}{N} \left(\frac{\epsilon_r - 1}{\epsilon_r + 2} \right).$$

This equation is known as the **Clausius-Mossotti** formula, or, in its application to optics, the **Lorentz-Lorenz** equation.